

POTENTIAL BIOGENETIC PRECURSORS OF WITHANOLIDES FROM *WITHANIA COAGULANS*

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Key Word Index—*Withania coagulans*; Solanaceae; withanolides; steroidal lactones; ergostene derivative.

Abstract—The structure of two new compounds, a withanolide and an ergostene derivative, are described and identified as 3 β ,14 α ,17 β ,20 α -tetrahydroxy-1-oxo-20S,22R-witha-5,24-dienolide and ergosta-5,25-diene-3 β ,24 ξ -diol. These are considered interesting intermediates in the biogenetic sequence leading to the formation of withanolides, the ergostene diol being a very early precursor.

INTRODUCTION

Within the framework of a collaborative effort oriented towards the study of the constituents of the fruits of *Withania coagulans* Dunal., a number of compounds were isolated, three of which were identified as sitosterol, 3 β -hydroxy-2,3-dihydro-withanolide F (1a), and ergosta-5,25-diene-3 β ,24 ξ -diol (3). The latter two, which are new compounds, can be considered as intermediate products in the biosynthetic sequence which produces the withanolides.

RESULTS

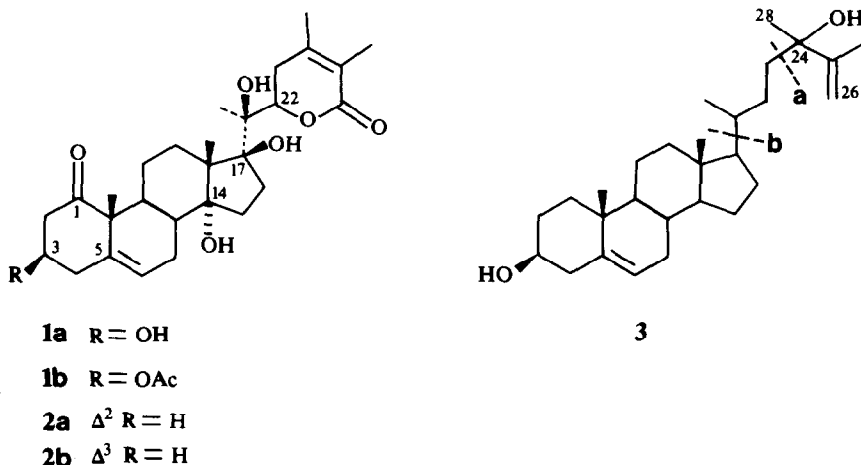
Compound 1a, C₂₈H₄₀O₇, showed in the UV spectrum

an absorption at λ_{\max} 224 nm (ϵ 10,800). This value and its intensity are characteristic for a single α,β -unsaturated carbonyl chromophore. The IR spectrum indicated a six-membered ring ketone, an unsaturated δ -lactone and a double bond (1700, 1670 and 1650 cm⁻¹). The ¹H NMR spectrum (Table 1) was characteristic in a general way for the steroidal structure of the withanolide class [1] and more specifically for the withanolide E group, i.e. having an α -oriented side chain [2, 3]. Indeed, the 21-Me signal is a singlet at δ 1.39, the 18-Me is at 1.09 and the 22-H shows a double-doublet at 4.88, all values which are close to withanolide E and certain related compounds. Interestingly, in the low field region of the spectrum, no signals could be observed for the usual 2-en-1-one system of

Table 1. ¹H NMR spectral signals of relevant protons in compounds 1–3

Compound	2-H	3-H	4-H	6-H	22-H	26-H	Methyl groups				
							18	19	21	27	28
1a	—	3.92 (br octet)	—	5.63 (m*7)	4.88 (dd 9, 7)	—	1.09 (s)	1.26 (s)	1.39 (s)	1.86 (s)	1.94 (s)
		5.03 (br octet)		5.67 (m*7)	4.89 (dd 9, 7)		1.12 (s)	1.24 (s)	1.43 (s)	1.88 (s)	1.94 (s)
1b	—	5.85 (dd 10, 2)	6.76 (ddd 10, 5, 2)	2.82 (dd 19, 6)	5.60 (m*7)	4.93 (dd, 9, 7)	1.13 (s)	1.23 (s)	1.42 (s)	1.88 (s)	1.94 (s)
				3.27 (br d 19)							
2a	—	5.68 (br d 10)	6.05 (br d 10)	5.70 (m*10)	4.91 (dd 9, 7)	—	1.13 (s)	1.37 (s)	1.43 (s)	1.87 (s)	1.93 (s)
		3.29		5.34 (d 5)		4.81 (s*5)	0.67 (s)	1.00 (s)	0.92 (d 6.6)	1.73 (s)	1.30 (s)
3	—					4.95 (s*4)					
		[3.85]		[5.43]		[5.44] [5.75]	[0.67]	[1.06]	[1.02]	[1.94]	[1.56]

Chemical shifts are as δ values; coupling constants (in Hz) are in parentheses; data measured in C₅D₅N are in square brackets; * for $W_{1/2}$ values.



ring A of the withanolides, and only one olefinic proton was present at δ 5.63, which was assigned to 6-H for a Δ^5 -bond. A broad octet at δ 3.92 was related to a proton (3α -H) adjacent to the 3β -OH and was characteristic for such a group. All these data together with the location of the 19-Me (δ 1.26) were in close agreement with those of a semi-synthetic derivative, obtained from previous work, possessing a 1-oxo- 3β -OH- Δ^5 system [4], and therefore structure **1a** was assigned to this compound. Upon acetylation, a mono- 3β -acetate (**1b**) was obtained, in which the proton geminal to the acetoxy group had moved, as expected, to δ 5.03.

The ^{13}C NMR data of compounds **1a** and **1b** are given in Table 2. All signals for the carbon atoms of rings C, D and the side chain have values similar to those of withanolide E and its derivatives [5]. The outstanding values, however, are for the specific substitutions of ring A in **1a** and **1b** respectively; they are related to a C-1 regular carbonyl, to the C-3 to which an alcohol (or an acetate) is attached, and to the C-5 and C-6 of the Δ^5 -bond. It is noteworthy that when converting the 3-OH to 3-OAc, the C-2 and C-4 became more shielded and showed upfield shifts of 4 and 2 ppm, respectively.

Further confirmation for the structure was adduced by

mass spectrometry. As for withanolide E, the molecular ion was absent, but fragments corresponding to the loss of 1–4 times H_2O were present. The cleavage of the C(20)–C(22) bond was observed through fragments indicating the loss of $\text{C}_7\text{H}_9\text{O}_2$, together with H_2O ; the cleavage of the C(17)–C(20) bond resulting in loss of $\text{C}_9\text{H}_{13}\text{O}_3$, and again minus H_2O , was observed as well. All these are characteristic fragmentations of the withanolide class of compounds.

Since compound **1a** was considered an intermediate in the biogenesis of the withanolides, it was further subjected to a biomimetic reaction, namely the elimination of the 3β -OH, in order to produce the characteristic 2-en-1-one system of ring A of the withanolides. To this end, elimination of the 3-acetoxy group in **1b** was carried out with alkali in dioxane, as described earlier with a semi-synthetic model [4]. In the present case, withanolide F (**2a**) was obtained as a major product and was found to be in all respects identical with an authentic sample. This compound was obtained together with its Δ^3 -isomer, namely Δ^3 -isowithanolide F (**2b**). The ^1H NMR and ^{13}C NMR spectra of compound **2b** show the characteristic signals for the protons at C-3, C-4 and C-6, and for C-3 to C-6 reported under different circumstances [5, 6].

Table 2. ^{13}C NMR spectral data of compounds **1a**, **1b** and **2b**

Carbon	1a	1b	2b	Carbon	1a	1b	2b
1	209.7	211.3	210.5	16	37.1	37.1	37.9
2	47.6	43.6	39.7	17	87.9	88.1	87.9
3	68.6	70.2	127.9	18	20.6	20.6	20.6
4	40.0	38.0	129.5	19	18.4	17.3	20.5
5	135.4	134.9	140.5	20	78.7	79.2	79.1
6	125.9	126.3	121.2	21	19.1	20.0	19.9
7	25.9	26.0	25.9	22	81.5	80.0	79.8
8	36.2	36.1	33.9	23	32.5	32.4	32.4
9	35.9	36.1	36.1	24	152.3	150.4	150.4
10	53.1	53.0	52.3	25	121.4	121.7	121.5
11	22.2	22.0	21.8	26	166.0	165.8	165.9
12	34.6	34.3	34.2	27	12.4	12.4	12.4
13	54.1	54.9	53.8	28	20.7	20.6	20.6
14	82.5	82.0	83.1	CH ₃ -CO	—	21.1	—
15	30.4	30.4	30.1	CH ₃ -CO	—	170.1	—

Compound **3** was found to be a sterol, $C_{28}H_{46}O_2$, showing in the 1H NMR the characteristic values for a 3β -hydroxy-5-ene system (Table 1). Two low-field, sharp one-proton signals (δ 4.81 and 4.95), together with a signal at 1.73 for a methyl group on a double bond, were assigned to the terminal methylene group shown in structure **3**. By comparing the values of the two methylenic protons, which are 0.14 ppm apart, with those of similar systems [7] in 2-methyl-2-propenol (δ 4.8 and 4.9) and in 3-methyl-3-butenol (δ 4.75 for two protons), it was deduced that in the present case a hydroxy group could be adjacent as in the first example, and therefore located at C-24. Indeed, the 28-Me attached at C-24 (as in ergostane) appeared as a singlet at δ 1.30, forming a side chain as drawn in **3**. This assumption was strongly supported by measuring the 1H NMR spectral shifts in C_5D_5N for this compound. Due to the collision complex formed with the 24-OH, the 27 and 28-Me exhibited a shift of $\Delta_{C_5D_5N}^{CDCl_3}$ 0.21 and 0.26 ppm, respectively, whereas an even stronger shift was measured for the 26-methylenic protons (0.63 and 0.80 ppm). All these data indicate the proximity of the tertiary hydroxy group to the methylene group, which was therefore placed at C-25.

Further support for the side chain as drawn in **3** was adduced through the high-resolution mass spectrum. Two cleavages *a* and *b* could be well visualized. The first one provided C_5H_9O (85) fraction of the molecule which gave the 100% peak. Also present were $M - 85 = 329$ and $M - 18 - 85$. The fraction C_4H_8 (56) for cleavage *b* following *a* was also well represented. All peaks for the carbocyclic skeleton minus cleavage *a* and *b* with loss of the elements of water were present, as described in the Experimental.

DISCUSSION

In a series of preceding papers [8–10] dealing with the study of withanolides in different hybrids obtained by selected cross-breeding, in addition to genetic studies at the chemical level, it was possible to account for biogenetic pathways based on the structure of the different components fitting into the proposed sequence [4, 11]. Generally speaking, studies on additional chemotypes or species may contribute and support the ideas presented at different occasions. For example, it has been previously suggested that ring A is produced following a sequence **a–d** shown in Scheme 1. Presently, the study of *Withania coagulans* and the isolation of compound **1a** from natural sources having the substitution shown in **c** and its conversion to **d** provide now the complete pattern, and thereby all the combinations shown in **a** to **d** have now been actually isolated from different withanolide-producing plants.

Along the same line, the identification of the sterol **3** is of importance and may provide additional information

and the clue to the way whereby the lactonic side chain is formed in the plant. 24-Methylenecholesterol (**a** in Scheme 2) has been shown to be a precursor and at the origin of the δ -lactone side chain of the withanolide [12]. It is now proposed that compound **3** (**b** in Scheme 2) is actually one of the intermediate steps leading to the lactone. In such a case, **b** should undergo 22-hydroxylation to **c**, a well-known reaction in the steroidal pathway in nature [13]. From this point two alternative hypotheses may be considered as shown in the scheme: one going through cyclization (**d**) and oxidation (**e**); and the second through oxidation (**g**) of the allylic isomer (**f**), followed by cyclization to **e**. The latter compound, a lactol, has been found to be present in a number of withanolides [14–17]. Ultimate oxidation will lead to the δ -lactone **h**.

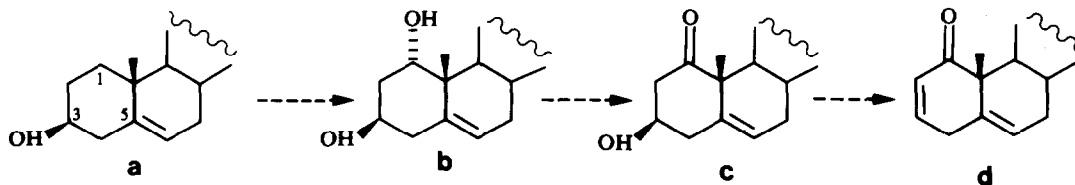
EXPERIMENTAL

Mps were measured on a Fischer–Johns apparatus and are uncorr. Optical rotations were determined in $CHCl_3$ solns. IR spectra were recorded with KBr pellets; UV spectra were recorded for EtOH solns; 1H NMR and ^{13}C NMR spectra were determined on a Bruker WH 270 and WH 90 (22.63 MHz) instruments respectively, for solns in $CDCl_3$, containing TMS as internal standard. Liquid column chromatography was done over Si gel H or G 60 and over Al_2O_3 . Analytical TLC was carried out using chromatoplates (50 \times 75 mm, Si gel F_{254}), and prep. TLC using chromatoplates, 200 \times 200 \times 2 mm, Si gel F_{254} . Mass spectra were determined under the direction of Dr. Z. Zaretskii, with a Varian MAT 731 for HR, and an improved Atlas CH4 for LR.

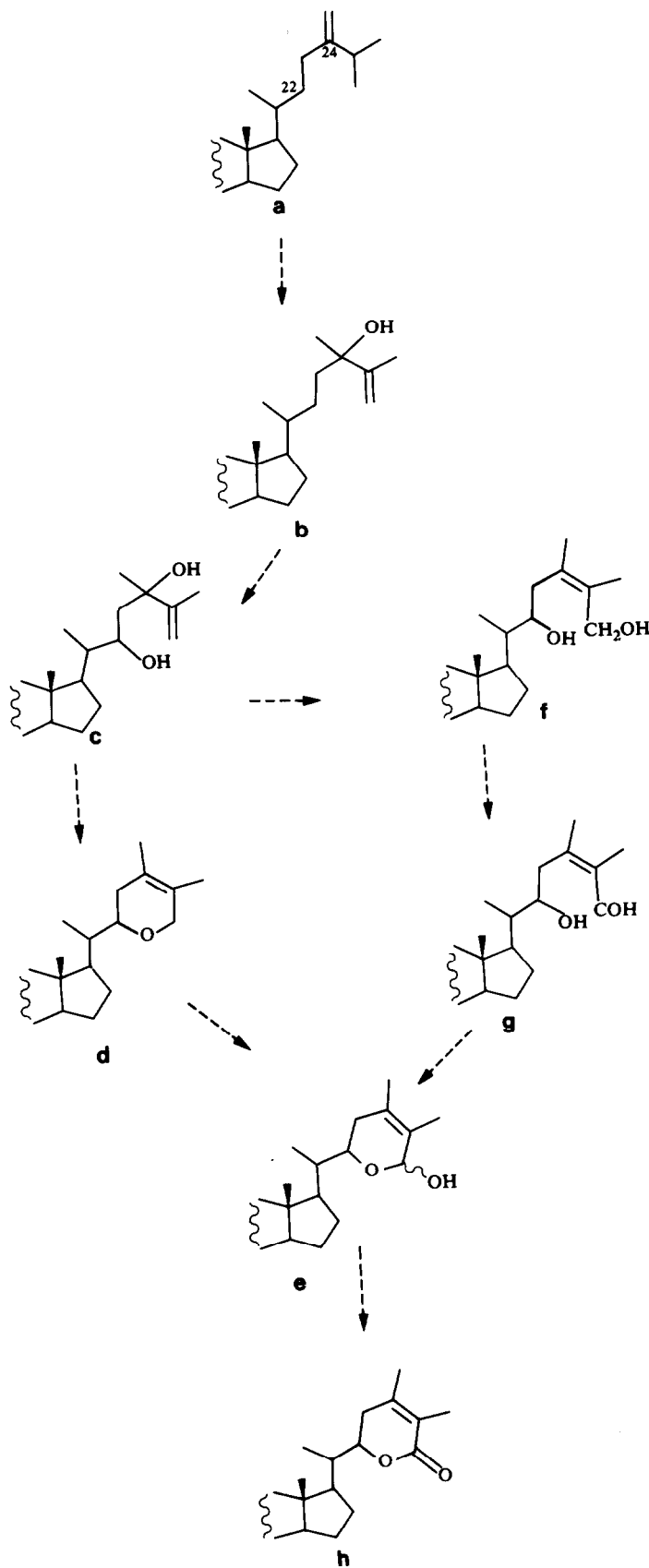
Isolation of compounds 1a and 3. Air-dried powdered fruits of *Withania coagulans* Dunal. (authenticated by the Director, Regional Research Laboratory, Jammu Tawi) were extracted with H_2O by repeated maceration; this fraction was extracted with $CHCl_3$, washed with H_2O , dried (Na_2SO_4), and evaporated at red. pres. The residue was chromatographed over Si gel H, using successively petrol (bp 60–80°), C_6H_6 , $CHCl_3$, EtOAc and MeOH as eluants. The first three yielded fatty material, whereas from EtOAc a crystalline product was obtained, which was purified by repeated crystallizations from MeOH– Me_2CO to yield compound **1a**.

The petrol extract of the fruits was saponified and the unsaponifiable fraction extracted with Et_2O . The residue following evaporation was chromatographed over Al_2O_3 and eluted successively with petrol, C_6H_6 , $CHCl_3$ and MeOH in various combinations. The fraction eluted with C_6H_6 – $CHCl_3$ (1:1) gave compound **3**, which crystallized from $CHCl_3$. Further elution with the same solvent mixture gave a compound which crystallized from Me_2CO , mp 137–138°, identified as sitosterol by comparison with an authentic sample, similar 1H NMR and mmp.

20S, 22R-3 β , 14 α , 17 β , 20 α -Tetrahydroxy-1-oxo-witha-5, 24-dienolide (**1a**). Mp 260–261° (from MeOH– Me_2CO); $[\alpha]_D^{25} + 61.4^\circ$ (c 0.2); UV λ_{max} nm: 224 (ϵ 10 800); IR ν_{max} cm^{-1} : 3400, 1700,



Scheme 1.



Scheme 2.

1670 and 1650; MS (LR) m/z (rel. int.): 452 (3) $[M - 2 \times H_2O]^+$, 434 $[M - 3 \times H_2O]^+$ (4), 416 $[M - 4 \times H_2O]^+$ (1), 345 $[M - 125 - H_2O]^+$ (9), 327 $[M - 125 - H_2O]^+$ (9), 309 $[M - 125 - 3 \times H_2O]^+$ (6), 283 $[M - 169 - 2 \times H_2O]^+$ (17), 125 [cleavage C20–C22] $^+$ (62), 169 [cleavage C17–C20] $^+$ (32). MS (HR) m/z (rel. int.): 434.2446 $[C_{28}H_{34}O_4]^+$ (10), 345.2065 $[C_{21}H_{29}O_4]^+$ (14), 327.1931 $[C_{21}H_{27}O_3]^+$ (16), 309.1847 $[C_{21}H_{25}O_2]^+$ (11), 301.1788 $[C_{19}H_{25}O_3, M - 169 - H_2O]^+$ (36), 283.1696 $[C_{19}H_{23}O_2]^+$ (23), 255.1774 $[C_{18}H_{23}O, M - 169 - 2 \times H_2O - CO]^+$ (13), 241.1600 $[C_{17}H_{21}O, M - 125 - 2 \times H_2O - \text{cleavage of ring A}]^+$ (14), 197.1318 $[C_{15}H_{17}, M - 169 - 2 \times H_2O - \text{cleavage of ring A}]^+$ (21), 125.0595 $[C_7H_9O_2]^+$ (89), 169.0874 $[C_9H_{13}O_3]^+$ (42). (Found: C, 68.70; H, 8.16. $C_{28}H_{40}O_7$ requires: C, 68.85; H, 8.20%, MW 488.6.)

20S,22R-3 β -Acetoxy-14 α ,17 β -20 α_F -trihydroxy-1-oxo-witha-5,24-dienolide (**1b**). Acetylation was carried out with Ac₂O–pyridine at room temp. for 24 hr. Cryst. from EtOAc, mp 154–155°; $[\alpha]_D^{25} + 33^\circ$ (c 0.15); UV λ_{\max} nm: 224 (ϵ 16000); IR ν_{\max} cm⁻¹: 3400, 1735, 1705, 1670, 1650 and 1230; MS (LR) m/z (rel. int.): 512 $[M - 18]^+$ (1), 452 $[M - 60 - 18]^+$ (5), 434 $[M - 60 - 2 \times 18]^+$ (2), 387 $[M - 125 - 18]^+$ (3), 369 $[M - 125 - 2 \times 18]^+$ (2), 327 $[M - 60 - 18 - 125]^+$ (7), 309 $[M - 60 - 125 - 2 \times 18]^+$ (12), 169 [cleavage of C17–C20, $C_9H_{13}O_3$] $^+$ (60), 125 [cleavage C20–C22, $C_7H_9O_2$] $^+$ (38). (Found: C, 67.98; H, 7.99. $C_{30}H_{42}O_8$ requires: C, 67.51; H, 7.82%. MW 530.6.)

20S,22R-14 α ,17 β -20 α_F -Trihydroxy-1-oxo-witha-2,5,24-trienolide (withanolide F) (**2a**). The acetate **1b** (35 mg) was treated for 15 min in 1 M KOH in dioxane and the reaction worked up as usual. A mixture of **2a** and **2b** was obtained which was resolved on chromatoplates. The lower spot (21 mg) was identified as withanolide F (**2a**), mp 191–192° (from EtOAc); UV λ_{\max} nm: 226 (ϵ 17000); IR ν_{\max} cm⁻¹: 1680; undepressed mmp with an authentic sample. The second compound could not be induced to crystallize and was identified only through its ¹H NMR spectrum as **2b** (isowithanolide F).

3 β ,24 ξ -Dihydroxy-ergosta-5,25-dienolide (**3**). Mp 208–209° (from CHCl₃); $[\alpha]_D^{25} + 68.9^\circ$ (c 0.14); UV λ_{\max} nm: 210 sh (ϵ 7600); IR ν_{\max} cm⁻¹: 3300, 3080 ($R_2C=CH_2$ stretching), 1675, 1625 ($C=C \Delta^{5,25}$), and 895 ($C-H$ out of plane deformation). MS (LR) m/z (rel. int.): 414 $[M]^+$ (2), 396 $[M - 18]^+$ (12), 329 (2), 311 (1), 255 (10), 85 (91), 56 [cleavage *a* and *b* C_4H_8] $^+$ (15), 57 $[C_4H_8 + H]^+$ (90). MS (HR) m/z (rel. int.): 414.3536 $[M]^+$ (1), 329.2848

[cleavage *a* $C_{23}H_{37}O$] $^+$ (3), 311.2749 $[M - 18 - 85 C_{23}H_{35}]^+$ (1), 255.2101 $[M - 18 - 141 C_{19}H_{27}]^+$ (12.5), 85.0671 [cleavage *a* C_3H_9O] $^+$ (100), 141.1245 [cleavage *b* $C_9H_{17}O$] $^+$ (1), $C_{28}H_{46}O_2$, MW 414.65.

REFERENCES

1. Glotter, E., Kirson, I., Lavie, D. and Abraham, A. (1978) *Bioorganic Chemistry* (van Tamelen, E. E., ed.), Vol. 2, p. 57. Academic Press, New York.
2. Lavie, D., Kirson, I., Glotter, E., Rabinovich, D. and Shakked, Z. (1972) *J. Chem. Soc. Chem. Commun.* 877.
3. Glotter, E., Abraham, A., Günzberg, G. and Kirson, I. (1977) *J. Chem. Soc. Perkin Trans 1*, 341.
4. Vande Velde, V. and Lavie, D. (1981) *Phytochemistry* **20**, 1359.
5. Gottlieb, H. E. and Kirson, I. (1981) *Org. Magn. Reson.* **16**, 20.
6. Kirson, I. and Gottlieb, H. E. (1980) *J. Chem. Res. S*, 388; M, 4275.
7. Pouchert, C. J. and Campbell, J. R. (1974) *The Aldrich Library of NMR Spectra*, Vol. 1, p. 101. Aldrich Chemical Company, Milwaukee, Wisconsin.
8. Kirson, I., Abraham, A. and Lavie, D. (1977) *Isr. J. Chem.* **16**, 20.
9. Eastwood, F. W., Kirson, I., Lavie, D. and Abraham, A. (1980) *Phytochemistry* **19**, 1503.
10. Nittala, S. S. and Lavie, D. (1981) *Phytochemistry* **20**, 2741.
11. Vande Velde, V. and Lavie, D. (1982) *Phytochemistry* **21**, 731.
12. Lockley, W. J. S., Roberts, D. P., Rees, H. H. and Goodwin, T. W. (1974) *Tetrahedron Letters* 3773.
13. Glotter, E., Kirson, I., Lavie, D. and Abraham, A. (1978) *Bioorganic Chemistry* (van Tamelen, E. E., ed.), Vol. 2, pp. 89–90 and refs. cited therein. Academic Press, New York.
14. Begley, M. J., Crombie, L., Ham, P. J. and Whiting, D. A. (1972) *J. Chem. Soc. Chem. Commun.* 1108.
15. Bates, R. B. and Morehead, S. R. (1974) *J. Chem. Soc. Chem. Commun.* 125.
16. Begley, M. J., Crombie, L., Ham, P. J. and Whiting, D. A. (1973) *J. Chem. Soc. Chem. Commun.* 821.
17. Kirson, I., Gottlieb, H. E. and Glotter, E. (1980) *J. Chem. Res. S*, 125; M, 2134.